

4-(4-Methoxyphenyl)-3-methyl-5-(2-pyridyl)-  
4H-1,2,4-triazole dihydrateZuo-Xiang Wang,\* Yan Lan,  
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## Key indicators

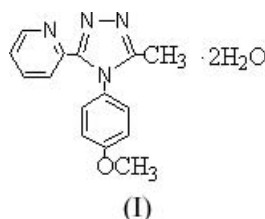
Single-crystal X-ray study  
 $T = 295$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.002$  Å  
 $R$  factor = 0.048  
 $wR$  factor = 0.115  
Data-to-parameter ratio = 15.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.

In the title compound,  $\text{C}_{15}\text{H}_{14}\text{N}_4\text{O}\cdot 2\text{H}_2\text{O}$ , the 1,2,4-triazole, pyridine and benzene rings do not share a common plane. The crystal structure is stabilized by  $\text{O}-\text{H}\cdots\text{N}$  and  $\text{O}-\text{H}\cdots\text{O}$  intermolecular hydrogen-bond interactions.

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## Comment

Substituted 1,2,4-triazoles have attracted considerable attention in recent years (Bencini *et al.*, 1987; Koningsbruggen *et al.*, 1997; Moliner *et al.*, 1998; Klingele & Brooker, 2003). This is mainly because of the fact that their ligand strength is in the appropriate region to give spin-crossover complexes with iron(II) salts, which could be used as molecular-based memory devices, displays and optical switches (Garcia *et al.*, 1997; Kahn & Martinez, 1998). We have synthesized a new compound, 4-(*p*-methoxyphenyl)-3-methyl-5-(2-pyridyl)-1,2,4-triazole, and we report here the crystal structure of its dihydrate, (I).



The three rings of (I) do not share a common plane. The dihedral angle between the 1,2,4-triazole and pyridine rings is  $38.22$  ( $9$ ) $^\circ$ , and that between the 1,2,4-triazole and substituted benzene rings is  $82.25$  ( $6$ ) $^\circ$ .

Two N atoms of the 1,2,4-triazole ring form intermolecular hydrogen bonds with the water molecules (Table 2). An additional strong hydrogen-bond interaction involving the water molecules is observed.

## Experimental

The title compound was synthesized by the reaction of 4,4'-dimethoxyphenylphosphazone with *N*-acetyl-*N'*-(2-pyridyl)hydrazine in *o*-dichlorobenzene at 463–473 K (Grimmel *et al.*, 1946; Klingsberg, 1958). Single crystals suitable for X-ray diffraction were obtained by recrystallization from water.

## Crystal data

$\text{C}_{15}\text{H}_{14}\text{N}_4\text{O}\cdot 2\text{H}_2\text{O}$   
 $M_r = 302.33$   
Monoclinic,  $P2_1/c$   
 $a = 11.629$  (3) Å  
 $b = 9.927$  (2) Å  
 $c = 14.019$  (3) Å  
 $\beta = 106.897$  (4) $^\circ$   
 $V = 1548.5$  (6) Å $^3$   
 $Z = 4$

$D_x = 1.297$  Mg m $^{-3}$   
Mo  $K\alpha$  radiation  
Cell parameters from 793  
reflections  
 $\theta = 3.1$ – $25.1$  $^\circ$   
 $\mu = 0.09$  mm $^{-1}$   
 $T = 295$  (2) K  
Block, yellow  
 $0.31 \times 0.22 \times 0.20$  mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer  
 $\varphi$  and  $\omega$  scans  
 Absorption correction: multi-scan (SADABS; Bruker, 2000)  
 $T_{\min} = 0.97$ ,  $T_{\max} = 0.98$   
 7967 measured reflections

3038 independent reflections  
 2217 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.027$   
 $\theta_{\text{max}} = 26.0^\circ$   
 $h = -14 \rightarrow 14$   
 $k = -6 \rightarrow 12$   
 $l = -17 \rightarrow 16$

Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.048$   
 $wR(F^2) = 0.115$   
 $S = 0.97$   
 3038 reflections  
 201 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0683P)^2 + 0.0126P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.16 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.13 \text{ e } \text{\AA}^{-3}$

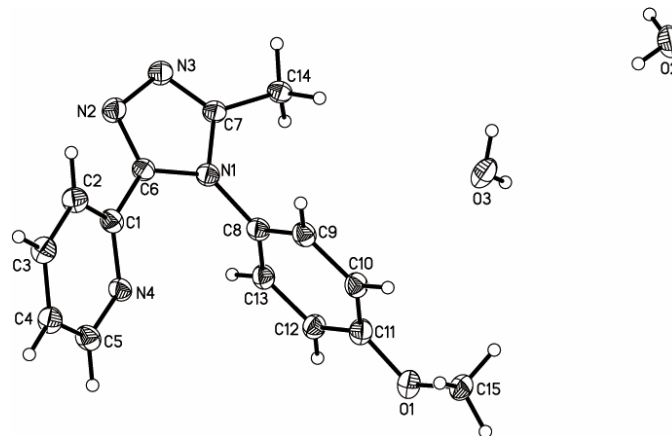


Figure 1 The molecular structure of the title compound, with the atom-labelling scheme. Displacement ellipsoids are shown at the 67% probability level.

Table 1

Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ ).

C6–N2	1.295 (2)	C11–O1	1.361 (2)
C6–N1	1.379 (2)	C15–O1	1.414 (2)
C7–N3	1.278 (2)	N2–N3	1.396 (2)
C7–N1	1.374 (2)		
N2–C6–N1	110.48 (14)	C7–N1–C6	103.61 (14)
N2–C6–C1	122.70 (15)	C7–N1–C8	126.22 (14)
N1–C6–C1	126.79 (15)	C6–N1–C8	130.10 (14)
N3–C7–N1	111.19 (15)	C6–N2–N3	107.09 (14)
N3–C7–C14	126.65 (15)	C7–N3–N2	107.60 (14)
N1–C7–C14	122.04 (15)		

Table 2

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
O2–H2D $\cdots$ N3 <sup>i</sup>	0.85	2.43	2.906 (2)	116
O3–H3B $\cdots$ N2 <sup>i</sup>	0.85	2.29	2.848 (2)	124
O2–H2C $\cdots$ O3 <sup>ii</sup>	0.85	2.06	2.783 (2)	143

Symmetry codes: (i)  $x, y - 1, z$ ; (ii)  $x, -y + \frac{1}{2}, z + \frac{1}{2}$ .

All H atoms were located in a difference Fourier map and allowed to ride on their parent atoms at distances of 0.85 (O–H), 0.93 (C–H aromatic) and 0.96  $\text{\AA}$  (C–H methyl), with  $U_{\text{iso}}(\text{H})$  values of 1.2–1.5 times  $U_{\text{eq}}$  of the parent atoms.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2000); software used to prepare material for publication: SHELXTL.

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